

1 ATSC 201 Fall 2019 Excel Tutorial

2 Tutorial Outline:

3 -Basic Introduction to Excel Terminology and Functions

4 -Example problems using equations and variables

5 -Format of Homework

6 7 **Excel Terminology:**

8 -Worksheet (spreadsheet): Your default display within excel containing a bunch of grid boxes
9 called 'cells.'

10
11 -Cell: An individual box on a worksheet. A cell can contain three types of items: (1) text
12 characters (ie words or sentences) , (2) numerical values, and (3) formulas. Each cell is
13 identified by a column letter and row number. For example, the cell in the top left of the
14 worksheet is called cell 'A1.' The cell immediately to the right is 'B1.' Underneath A1 is A2.

15 16 *Using Cells:*

17 Click in a cell and start typing words, numbers, or a formula. Let's start with text (no numbers).
18 Click on cell 'A1' and type in your name. When you are done typing, hit enter. Note that your
19 name may appear to also show up under column B if it is long enough but that the contents are
20 still all contained within cell A1. Numerical values can be entered in the same manner (type in a
21 value and hit enter).

22 23 *Using cells as a means of calculation (inputting a formula):*

24 Let's say we want to calculate the expression $579 + 621$. Excel can do all this for you when the
25 expression is entered correctly within one individual cell. Click on cell 'B3,' it is where we will
26 do this calculation. We cannot just type in '579+621' and hit enter and expect Excel to solve the
27 problem. Instead, for excel to know that this is a calculation, we must start with an '=', so within
28 B3 should be typed '=579+621', then hit enter. Note that what we have entered has disappeared
29 and what is left is the result, the answer!

30
31 Excel can handle more complex calculations as well involving sines, cosines, absolute values
32 and exponents just to name a few. Note, however that angles must be given in **Radians, not**
33 **Degrees**. If we wanted to find the cosine of an angle, say π , we would type '=cos(pi())'. Try it
34 in an available cell. You should obtain the answer '-1.'

35
36 The table on the back of this sheet provides a partial list of Excel mathematical functions.

Mathematical expression	In Excel	Notes
+	+	
-	-	
x	*	
÷	/	
a	abs(a)	Absolute value
a ^b	a^b	
sin(a)	sin(a)	Argument in radians
cos(a)	cos(a)	
tan(a)	tan(a)	
arcsin(a)	asin(a)	Returns in radians
arccos(a)	acos(a)	
arctan(a)	atan(a)	
π	pi()	No argument needed
e ^b	exp(b)	To get value of e, use exp(1).
ln(b)	ln(b)	Natural log
log(b)	log(b)	Base 10
log ₂ (b)	log(b,2)	Base 2 (2 can be replaced by other bases)
1x10 ⁻³	1E-3	E means “times ten to the power of”
(a) ^{1/2}	sqrt(a)	square root

41

42 **Defining and using variables:**

43 To learn how to define variables we will set up a sample problem. Our problem is:

44

45 “Plot the change in density of dry air with height in an isothermal (constant temperature)
 46 atmosphere. Use pressure (P) coordinates as a surrogate for height, with the highest pressure at
 47 the bottom of the vertical (y) axis. Plot for pressures of 100 - 0 kPa, at 10 kPa increments.

48 a) use T = 25 degC, and plot on a linear-linear graph

49 b) copy the graph from (a) and change the vertical (P) axis to be logarithmic. (i.e., semi-log
 50 graph)

51 c) copy the graph from (b) and use it to plot multiple semi-log graph curves. In addition to T =
 52 25 degC already plotted, add the curves for T = 0 degC and T = -25 degC.”

53

54 First, we need to set up the spreadsheet into an ideal printing format. Select Layout > Page
 55 Layout. This way, you can see how many columns will fit in one page.

56

57 When you have created all your answers, check all your pages to ensure that no lines extend into
 58 adjacent pages. Wrap your text or move your graphs as needed to avoid splitting the graph once
 59 you are ready to print the worksheet.

60 After you finished creating all your answers, save the excel spreadsheet as an xls or xlsx file, and
 61 save it again as a pdf file to print.

62 Now back to our exercise...

63
64 The Equation of State we will use to solve this problem is the Ideal Gas Law (eqn 1.18)

65
66 $P = \rho * R_d * T$

67
68 Where P is the total atmospheric pressure, ρ the dry air density, R_d the gas constant and T the
69 temperature of the air in Kelvin.

70
71 The first thing we must do when starting a HW exercise is put our name, student number,
72 homework number, date, etc in the header. Also, turn on page numbers in the footer.

73
74 Next, write the Chapter number and the exercise number. Then copy the problem statement into
75 the excel worksheet. If the assignment is in our textbook, you can do a screen capture of the
76 exercise and paste or drag the resulting image into the spreadsheet. Otherwise, insert a text box
77 into the spreadsheet (click "insert" menu, then "text box", then use your cursor to click and drag
78 to put the box where you want it in the spreadsheet). For this exercise, we need to use the text
79 box method, and then click in the box and start typing the problem statement.

80
81 Now, underneath our problem statement, we define our known and unknown variables, and make
82 any assumptions if needed. Then, write the equation number that you plan to use. If possible,
83 make a screen capture of that equation and drag or paste it into your spreadsheet.

84 Your excel worksheet should look something like:

ATSC 201		HW #2		Your name	
today's date 2019				Your SN	
Chapter 1					
1) Plot the change in density of dry air with height in an isothermal (constant temperature) atmosphere. Use pressure (P) coordinates as a surrogate for height, with the highest pressure at the bottom of the vertical (y) axis. Plot for pressures of 100 - 0 kPa, at 10 kPa increments.					
a) use T = 25 degC, and plot on a linear-linear graph					
b) copy the graph from (a) and change the vertical (P) axis to be logarithmic. (i.e., semi-log graph)					
c) copy the graph from (b) and use it to plot multiple semi-log graph curves. In addition to T = 25 degC already plotted, add the curves for T = 0 degC & T = -25 degC.					
Given:	P range =	100 to 0	kPa		
	Dry air				
	T =	25 degC	for (a)		
Find:	rho vs. P	?	kg/m3		
Assume:	Ideal gas law				
Solution:	Use eq. (1.18)	$P = \rho \cdot R_d \cdot T$			
	solve for rho:	$\rho = P / (R_d * T)$			
where	Rd =	0.287053 kPa·K ⁻¹ ·m ³ ·kg ⁻¹			

85

86 **Part (a) of the Exercise:**

87 Skip down a few rows, and then in column B type the column header "P (kPa)". Under that
 88 header, type 100 as the first pressure. Under that, type 90 as the next pressure. Then, we can use
 89 the automatic sequence completion feature of Excel to enter the other pressures for us. To do
 90 that, click once in the center of the cell holding the 100 value, and drag the cursor to the center of
 91 the cell holding 90. Then let up on the cursor -- those first two pressure cells should have a box
 92 around them. If you position the cursor of the bottom right corner of that box, then you can drag
 93 that corner (called a "handle") downward to extend the pressure values to 0 pressure. (see
 94 example below).
 95

96 Now, let's use variables to start solving the problem. Planning ahead, we know that we will
 97 eventually need to plot 3 curves, one for each of the 3 temperatures in part (c). We can lay out
 98 the spreadsheet solve for part (a) while planning ahead for parts (b) and (c). To do this, two cells
 99 above the "P (kPa)", type the words "T (degC)". In the cell below that, type "T (K)". The reason
 100 for this is that thermodynamic equations almost always need to use temperatures in kelvin, so we
 101 will need to convert the temperatures that were given.
 102

103 Just to the right of the cell holding "T (degC)", type the temperature given for part (a): 25 .
 104 Then, in the "T (K)" row under the 25 value, enter the formula to calculate the kelvin
 105 temperature. Don't forget to start with an equal sign: “=(cell ID for temperature in
 106 Celsius)+273.15”. After you hit "enter", the result of the calculation will appear in the cell. You
 107 can always go back and see the formula by clicker in the cell again.

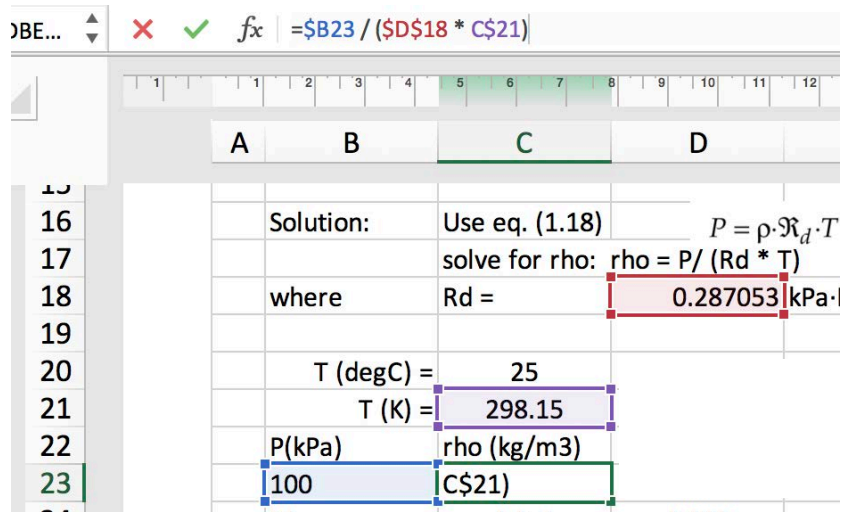
108 The bottom part of you sheet should look like:
 109

Solution:	Use eq. (1.18)	$P = \rho \cdot R_d \cdot T$	
	solve for rho:	$\rho = P / (R_d \cdot T)$	
where	$R_d =$	0.287053 kPa·K ⁻¹ ·m ³ ·kg ⁻¹	
	T (degC) =	25	
	T (K) =	298.15	
P(kPa)			
100			
90			
80			
70			
60			
50			
40			
30			
20			
10			
0			

111 If you click in the cell holding the kelvin temperature, it will make the terms and factors in your
 112 formula different colours, and will use those same colours to highlight the corresponding cells on
 113 your spreadsheet. This is very handy for debugging your spreadsheet, because you can see
 114 which cells were used in your formula.
 115

116 Next, we need to enter the formula to calculate the first density. In the cell to the right of the 100
 117 kPa value, click in it and start your formula with an equal sign.

118 = cell holding P / (cell holding Rd * cell holding T in kelvin) . Enter the appropriate cell
 119 addresses by clicking in those respective cells. On my spreadsheet, after I have entered the
 120 formula, I can click once in the "formula bar" above the ABCD column labels, to make the cells
 121 that I referenced show us in the same colours that are in the formula. See the example:
 122



123 After viewing those coloured cells, hit "enter" to exit that colour display. Check to see that the
 124 calculated result for density is reasonable; namely, compare to Table 1-5 in the textbook.
 125

126 Next, we can take advantage of an Excel shortcut to use that one equation that we typed in, and
 127 "fill it down" next to all the other pressures. To do this, click once in the center of the cell
 128 holding that first density, and you will see a box around it. Let up on the cursor, and then grab
 129 the handle at the bottom right corner of this box, and pull it down next to all the pressures. This
 130 action copies your one equation for density into all the other cells below it. That process is called
 131 "fill down". You can similarly "fill right", "fill up", etc.
 132

133 When you finished filling down, you will probably get garbage or errors showing in those other
 134 cells. This is because of the fact that Excel uses "RELATIVE" references in the formula that you
 135 type in.
 136

137 (Be sure to Save your work frequently.)
 138
 139

140 For example, using the figure above, my equation for density is entered into cell C23. It refers to
 141 the pressure one cell to its left, and to the temperature 2 cells above it, and to Rd which is one
 142 column to the right and 5 rows up. These are all relative to the location of the formula.

143
 144 When you filled that equation down to the next cell below, the equation in that cell is looking for
 145 numbers in the same RELATIVE locations. Namely: the pressure one cell to its left, and to the
 146 temperature 2 cells above it, and to Rd which is one column to the right and 5 rows up.

147
 148 Although this worked for pressure, it didn't work for T and Rd. We really wished that the
 149 formula referred to an ABSOLUTE reference for those variables. Namely, we want it to always
 150 look for T in cell C21 in my spreadsheet above, and for Rd in cell D18.

151
 152 To fix this, click once in the center of the first equation that you had types, for density at P = 100
 153 kPa. Double click to edit the formula, and put dollar signs "\$" in front of the column letter and in
 154 front of the row number for those two cells. Namely, your density formula should refer to T in
 155 cell \$C\$21, and to Rd in cell \$D\$18 (or to the corresponding cells in your own spreadsheet). Hit
 156 enter.

157
 158 Now, when you click once in the center of that first cell that you just fixed, when you fill it down
 159 again, all the densities should look reasonable, as shown below.

160

Solution:	Use eq. (1.18)	$P = \rho \cdot R_d \cdot T$
	solve for rho: rho = P / (Rd * T)	
where	Rd =	0.287053 kPa·K ⁻¹ ·m ³ ·kg ⁻¹
T (degC) =	25	
T (K) =	298.15	
P(kPa)	rho (kg/m ³)	
100	1.168	
90	1.052	
80	0.935	
70	0.818	
60	0.701	
50	0.584	
40	0.467	
30	0.351	
20	0.234	
10	0.117	
0	0.000	

161
 162
 163 To plot these data, click once in the center of the cell labeled P(kPa), and drag down and to the
 164 right until you get to the last density value of 0.000 . All of those rows in those two columns
 165 should be selected. Next, use the "Insert" tab in the Excel window to insert a Chart. Be sure to

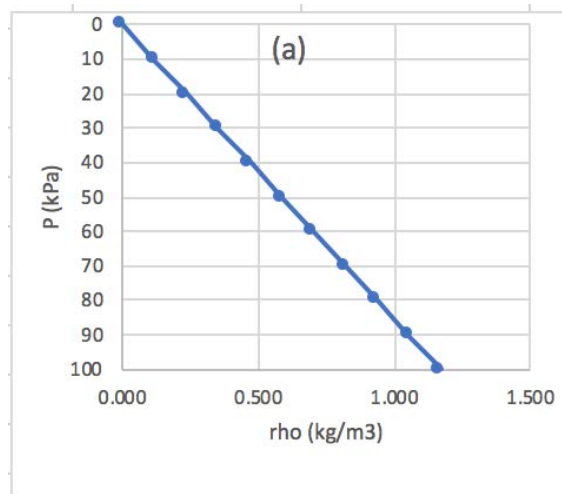
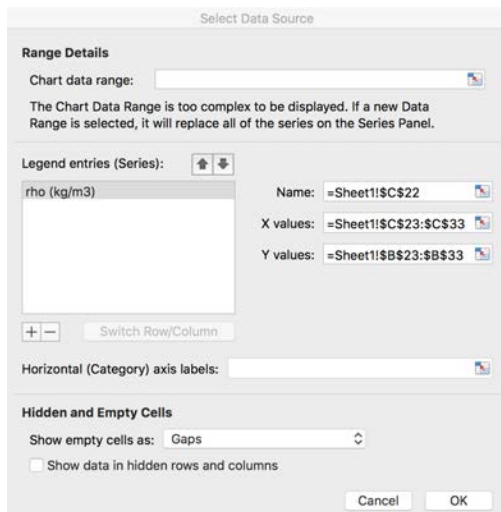
166 ALWAYS pick a scatter chart. Pick the scatter chart connected with straight lines. A chart will
167 appear on your spreadsheet -- perhaps you can drag it to page 2.

168
169 What it will show should hopefully be a straight line, with pressure values along the bottom axis
170 and density values along the vertical axis. But we want to switch the axes, and there is not a
171 simple button to do this in Excel (don't use the "Switch row/column" button -- it doesn't do what
172 we want). To make the switch, click once in the graph, which makes some new tabs appear at
173 the top of your Excel window. Select the "Chart Design" tab, then hit the "Select Data" button.

174
175 A pop-up window will appear that shows the range of X values in column B, and the range of Y
176 values in column C. But we want to switch the B's and C's. An easy way is to just type in this
177 change. Alternately, you can use the little icons to the right of the X and Y value lines to view
178 the original spreadsheet to select the appropriate column that you want for each axis. Regardless
179 of how you do it, after switching the axis the resulting "Select Data Source" box should look
180 something like the image at below left. Click the OK button.

181
182 Next, double click on the numbers in the vertical axis. This should cause a "Format Axis" box to
183 appear. In the Axis Options section, near the bottom, click on the check box for "Values in
184 reverse order". This will cause the high-pressure values to be at the bottom of the graph, to
185 correspond to the high pressures at the bottom of the atmosphere. Your graph should look like
186 the image below right (but without all the axis labels).

187



188

189

190 To make the axes numbers and labels look the way you want, first click once in the graph to get
191 the "Chart Design" tab to appear at the top of your window. Then select that tab, and then see
192 the "Add Chart Element" button at the left side of the chart design ribbon. When you click on it,
193 it drops down to allow you to change things like chart title, axes titles, etc. Use these to make
194 your graph look similar to the figure above right.

195

196 This completes part (a) of the exercise. Be sure to Save your work frequently.

197

198 **Part (b) of the Exercise:**

199

200 This part is easy. Click once in your graph from part (a) and copy it. Then click in an empty cell
201 to the right of that graph, and paste. (Or you can click once in the first graph, and then hold
202 down the option key (on a Mac) while you drag to an empty cell to the right.) The resulting
203 graph should look just like the first graph. This is a good starting point.

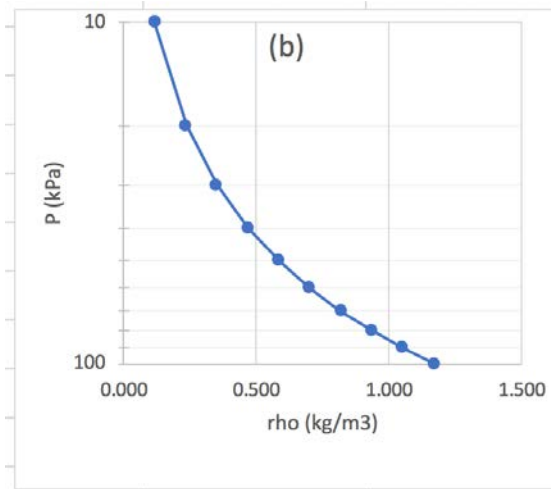
204

205 Next, in the new graph, double click on the numbers along the vertical axis. In the Format Axis
206 box that appear, near the bottom, click on the check box for "Logarithmic scale". It will give you
207 a warning about not being able to plot "0" on a log axis. Respond with "OK", then it should
208 produce the desired semi-log graph.

209

210 Click once on the chart, select the Chart Design tab, select Add chart element, and then add
211 minor grid lines for the log axis. Also change to chart title to say (b), and make any other labels
212 look nice. The result should look something like the figure below. Save your results.

213



214

215 Having achieved this is an important skill. We often plot Pressure in a reversed log axis in the
216 vertical, as a surrogate for height in the atmosphere. So you will use this skill a lot for future
217 homeworks.

218

219

220 **Part (c) of the Exercise:**

221

222 As a starting point, make a copy of your graph from part (b) and paste it in an empty part of page
223 2 of your spreadsheet.

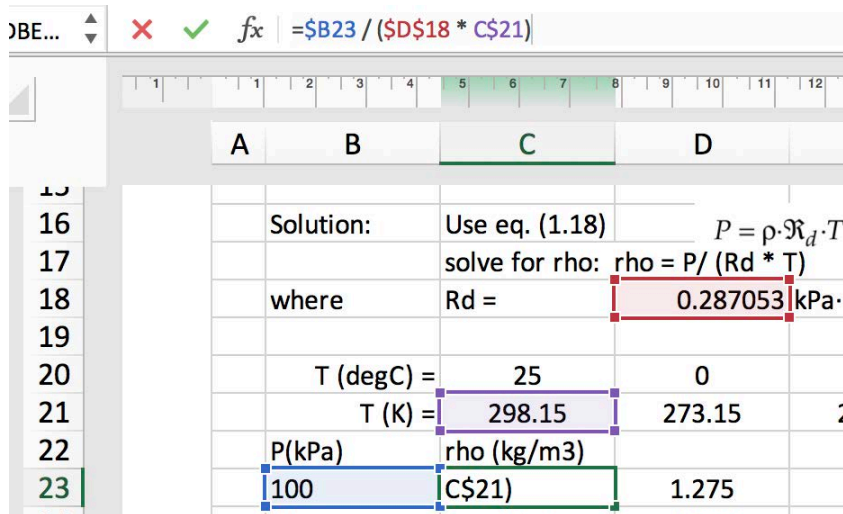
224

225 On your own: in the cells to the right of the 25 degC temperature, add the other temperatures for
 226 0 and -25 degC. Then use your new "fill" skills to take the formula for the first Kelvin
 227 temperature, and "fill right".
 228

229 Next, click once in the center cell holding the first density value at P=100 kPa and T = 25 degC.
 230 Then click on the handle in the bottom right of that one cell, and drag it to the right to "fill right".
 231 The result should give you garbage, or unrealistic numbers. As you might guess, this relates to
 232 RELATIVE vs ABSOLUTE references. Namely, we indeed wanted the cell for Rd to be an
 233 absolute reference, but for pressure we had wanted only the column to stay constant, so it always
 234 look in column B for the pressure. So go back to the formula in the first density cell to edit it.
 235 We want a dollar sign "\$" only in front of the column letter, not in front of the row number.
 236

237 Similarly, in that same formula, we want it to always look for temperature in that one row, but
 238 for different columns. So, in the first density formula, put a dollar sign in front of the row
 239 number, but remove it from in front of the column letter.
 240

241 The net result for that formula in the first density cell should look like
 242

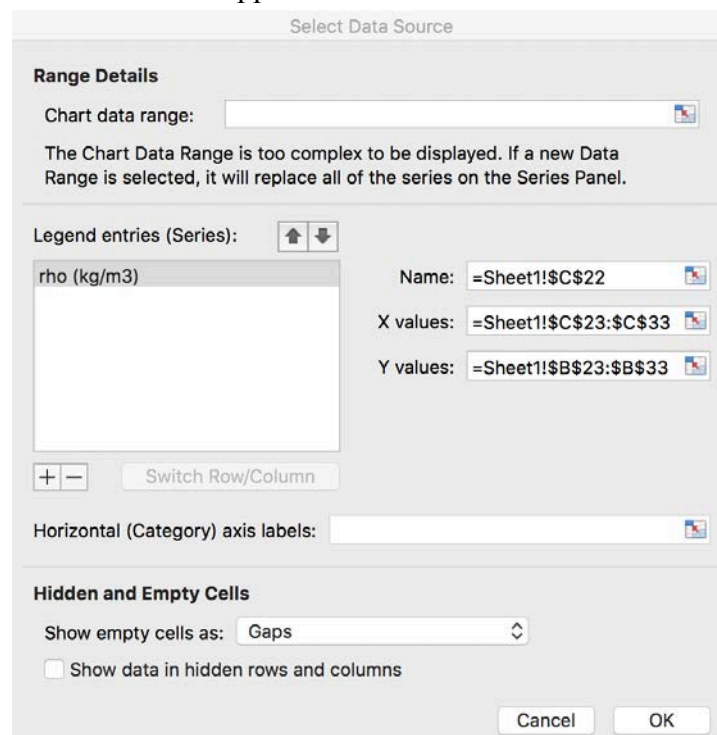


243
 244
 245 Then, after hitting enter, you can fill that cell down. Then you can select the whole column of
 246 densities, and fill right into the next two columns. The result should not have changed the
 247 numbers in the first density column, but should have produce the additional two columns as
 248 shown below.
 249
 250

Solution:	Use eq. (1.18)	$P = \rho \cdot R_d \cdot T$	
		solve for rho: $\rho = P / (R_d \cdot T)$	
where	$R_d =$	0.287053 kPa·K ⁻¹ ·m ³ ·kg ⁻¹	
T (degC) =	25	0	-25
T (K) =	298.15	273.15	248.15
P(kPa)	rho (kg/m ³)		
100	1.168	1.275	1.404
90	1.052	1.148	1.263
80	0.935	1.020	1.123
70	0.818	0.893	0.983
60	0.701	0.765	0.842
50	0.584	0.638	0.702
40	0.467	0.510	0.562
30	0.351	0.383	0.421
20	0.234	0.255	0.281
10	0.117	0.128	0.140
0	0.000	0.000	0.000

251
252
253
254
255

The final step in part (c) is to add these 2 new curves to the plot. To do this, click once on the new graph that you had copied from part (b), then select the Chart Design Tab, and finally select the Select Data button. The box that appears is the same as we saw before ...



256
257
258
259
260

... but now focus on the "+" button near the left edge. Click on it to add another curve to the graph, and then select the appropriate spreadsheet columns for the X and Y values. Do this again to add a third curve.

272 **Appendix. Summary of Plotting instructions using MS Excel (XY Scatter):**

273 Many HW assignments ask for plots to support your answer and most all of them are referring to
274 an XY Scatter type plot within Excel. Use these directions to guide your current and future
275 work.

276 These directions assume you already have plotable data sorted into columns.

277
278 **Step 1:** Click on a blank cell on your worksheet.

279
280 **Step 2:** At the top of the program click on the insert tab, then on the ‘scatter’ button and select the
281 appropriate type of scatter plot. I suggest ‘scatter with straight lines and markers.’. A blank frame should
282 appear so don’t worry, you are doing this correctly!

283
284 **Step 3:** Right-click on this blank frame and click on ‘select data.’

285
286 **Step 4:** A box titled ‘Select Data Source should appear with two columns: Legend Entries (Series) and
287 Horizontal (Category) Axis Labels. Under ‘Legend Entries’ click on the ‘Add’ button.

288
289 **Step 5:** A new box titled ‘Edit Series’ should pop up. It is from this dialogue where the data is selected
290 to place on your set of axes. We will do this step 3 times to produce 3 lines on one plot (Solar elevation
291 angles for 21-Dec, 23-Mar and 22-Jun).

292
293 **Step 6:** In the ‘series name’ box type in 21-Dec.

294
295 **Step 7:** In the ‘series X values’ box, click in the white area, then push the little button with the red,
296 upward pointing arrow immediately to the right.

297
298 **Step 8:** A new ‘edit series’ box will appear. It is here that we will select our x-coordinate data: the local
299 time of day. Highlight all of the local time data, not including the column title. Note Excel has inputted a
300 formula into the Edit series box for you. All that you need to do once this data is selected is push the
301 button in ‘edit series’ box with the downward red arrow. You should be brought back to the original ‘edit
302 series’ box.

303
304 **Step 9:** Repeat steps 7 and 8, except for the ‘Series Y values’ this time. You should delete the $=\{1\}$
305 before proceeding with selecting the data.

306
307 **Step 10:** Hit the ‘OK’ button on the main ‘edit series’ box and a graph will be generated. Note this is for
308 only the 21 December data. Start again with the ‘Add’ button from the Select Data Source box, which
309 should still be open, and follow steps 4-10 again for the other two cases. Then hit ‘OK’ on the Select
310 Data Source box and you will see your whole plot.

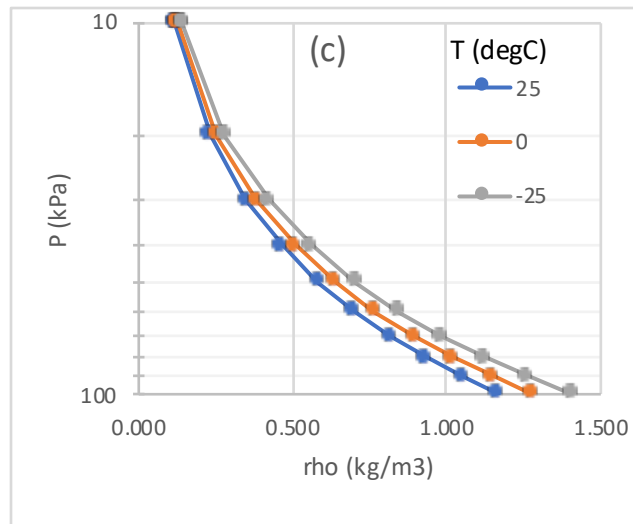
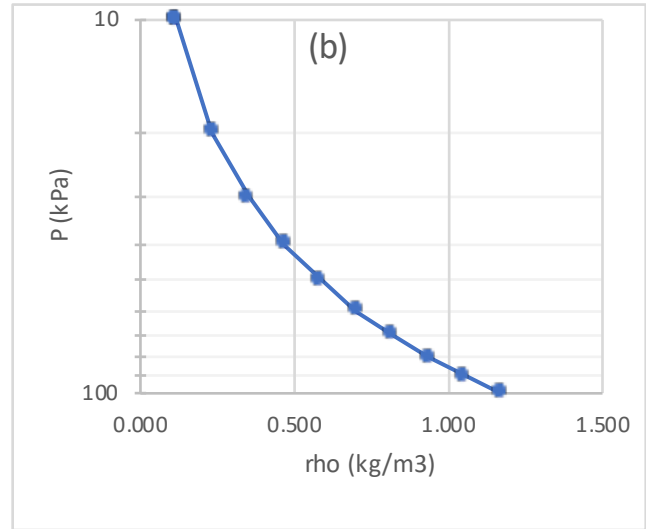
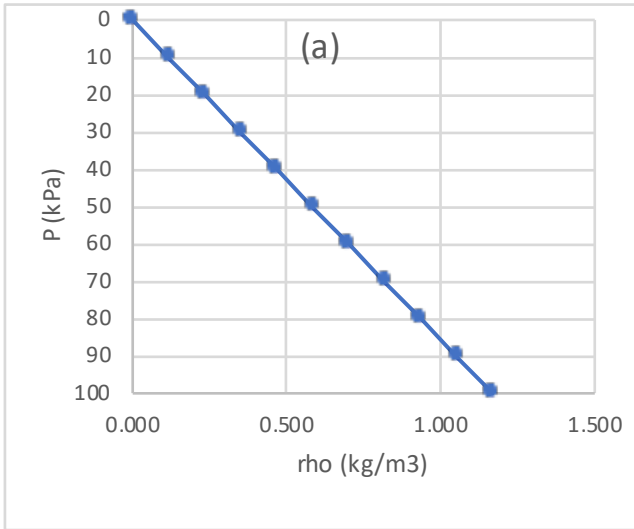
311
312 **All plots should be edited to have appropriate axis labels, units and legend. We will go over this**
313 **once everyone has reproduced the raw figure.**

- 1) Plot the change in density of dry air with height in an isothermal (constant temperature) atmosphere. Use pressure (P) coordinates as a surrogate for height, with the highest pressure at the bottom of the vertical axis. Plot for pressures of 100 - 1 kPa, at 10 kPa increments.
- a) use T = 25 degC, and plot on a linear-linear graph
- b) change the vertical (P) axis to be logarithmic. (i.e., semi-log graph)
- c) plot on the same semi-log graph multiple curves. In addition to T = 25 degC, also plot for T = 0 degC and T = -25 degC.

Given: P range = 100 to 0 kPa
 Dry air
 T = 25 degC for (a)
 Find: rho vs. P ? kg/m³
 Assume: Ideal gas law

Solution: Use eq. (1.18) $P = \rho \cdot R_d \cdot T$
 solve for rho: $\rho = P / (R_d \cdot T)$
 where $R_d = 0.287053 \text{ kPa} \cdot \text{K}^{-1} \cdot \text{m}^3 \cdot \text{kg}^{-1}$

P(kPa)	rho (kg/m ³)	rho (kg/m ³)	rho (kg/m ³)
	T (degC) = 25	T (degC) = 0	T (degC) = -25
	T (K) = 298.15	T (K) = 273.15	T (K) = 248.15
100	1.168	1.275	1.404
90	1.052	1.148	1.263
80	0.935	1.020	1.123
70	0.818	0.893	0.983
60	0.701	0.765	0.842
50	0.584	0.638	0.702
40	0.467	0.510	0.562
30	0.351	0.383	0.421
20	0.234	0.255	0.281
10	0.117	0.128	0.140
0	0.000	0.000	0.000



Check: Units OK. Curves approx. agree with Table 1-5.

Discussion: The air is thinner at higher altitudes (where the pressure is lower). So each breath of air would bring in fewer oxygen molecules at higher altitudes, causing me to become hypoxic.