ATSC 201 Fall 2025

Homework 1 Answer Key

Total mark out of 25

Chapter 1: A1g, A3g, A5g, A6g, A9g, A14g, A15g

Chapter 1

A1g) (4 marks)

Find the wind direction (degrees) and speed (m/s), given the (U,V) components: g) (-2,-10) mi/h

Given: U = -2 mi/h

V = -10 mi/h

Find: alpha (wind direction) M (wind speed)

Using eq. 1.1:

 $M = (U^2 + V^2)^0.5$

Using eq. 1.2a:

alpha = 90deg - (360deg/C)*arctan(V/U) + alpha0

C = 360 deg

if U < 0 alpha0 = 0 deg

M = 10.198039 mi/h = 4.55893137 m/s

M = 4.56 m/s alpha = 11.31 deg

Checks: Units ok. Physics ok.

Discussion: The wind is coming from the NNE with a magnitude of 4.56 m/s.

A3g)

Convert the following UTC time to local time in your own time zone: g) 18:00

(3 marks)

Given: 18:00 UTC

Find: local time

Using eqs. from Table 1-1

Vancouver's time zone is "U" (Pacific)

beta = 7:00 PDT alpha = 8:00 PST

Currently, local time is PDT.

DT = 11:00 PDT

for partial marks: ST = 10:00 PST

Checks: Units ok.

Discussion: Vancouver is currently in Pacific Daylight Time, so it is presently

7 hours behind UTC time.

A5g) (2.5 marks)

Find the pressure in kPa at the following heights above sea level, assuming an average T = 250K: g) 2 km

Given: z(m) = 2 km "=" 2000

avg T = 250 K

Find: P ? kPa

Using eq. 1.9a:

 $P = Po * e^{(-(a/T)*z)}$

where: Po = 101.3250 kPa

a = 0.0342 K/m

P = 77.1 kPa

Checks: Units ok. Physics ok.

Discussion:

At 2 km above sea level, pressure is about 76% of what it is at sea level. There are fewer air molecules above 2 km above sea level, than at sea level.

A6g) (3 marks)

Use the definition of pressure as a force per unit area, and consider a column of air that is above a horizontal area of 1 square meter. What is the mass of air in that column: g) Between pressure levels 100 and 50 kPa

Given: Pbottom = 100 kPa

Ptop = 50 kPa A = 1 m^2

Find: $\Delta m = ?$ kg

mass of air between Pbottom and Ptop

Using eq. 1.11:

 $\Delta m = (A/g)*(Pbottom - Ptop)$ where $g = 9.81 \text{ m/s}^2$

Convert Pbottom(kPa) and Ptop(kPa) to Pbottom(Pa) and Ptop(Pa):

Pbottom = 100000 Pa Ptop = 50000 Pa

 $\Delta m = 5096.84 \text{ kg}$

Checks: Units ok. Physics ok.

Discussion: This is a calculation of the air mass between 100 and 50 kPa pressure

that exists over 1 square meter of area.

A9g)

Convert the following temperatures: g) 2000°C = ?K

(2.5 marks)

Given: $T_c^{\circ} = 2000 \, ^{\circ}C$

Find: $T_K =$?

Using equation 1.7b:

$$T_K = T^{\circ}C + 273.15$$

T°F = 2273.15 K

Checks: Units ok.

Discussion: 2273.15K is hot enough to instantly melt pure iron

A14g) (4.5 marks)

What is the geopotential height and geopotential, given the geometric height? g) 5 km

Given: z = 5 km "=" 5000

Find: H = ? m

 $\Phi = \qquad ? \qquad \qquad m^2/s^2$

Using eq. 1.14a or b:

H = Ro*z(Ro+z)

where the radius of the Earth, Ro = 6356.766km = 6,356,766m.

Ro = 6356766 m

And using eq. 1.15:

 $\Phi = g^*H$ where $g = 9.81 \text{ m/s}^2$

H = 4996.07 m Φ = 49011.45 m^2/s^2

Checks: Units ok. Physics ok.

Discussion: The difference between the geometric height and

the geopotential height is relatively small at 5 km.

A15g)

(5.5 marks)

What is the standard atmospheric temperature, pressure, and density at the following geopotential height: g) 40 km?

Given: H = 40 km

Find: T = ? degC P = ? kPa

 $\rho =$? kg/m^3

Using eqs. 1.16 for $32km \le H \le 47km$:

T = 228.65 + 2.8*(H-32)

T = 251.05 K -22.10 degC

Using eq. 1.17 for $32km \le H \le 47km$:

 $P = 0.868*(228.65/T)^{12.2011}$

P = 0.28 kPa

Using and rearranging eq. 1.18:

 $\rho = P / Rd*T$ where Rd = 0.287053 kPa*m^3 / K*kg

gas constant for dry air

ρ = 0.00385094 kg/m^3

Checks: Units ok. Physics ok.

Discussion:

At 40 km, the pressure and density of the air is much less than at the surface